# Atmospheric Aerosol Physics, Physical Measurements, and Sampling

### Particle Mass Concentration Measurements

SAMLAC San Juan, Puerto Rico November 2018







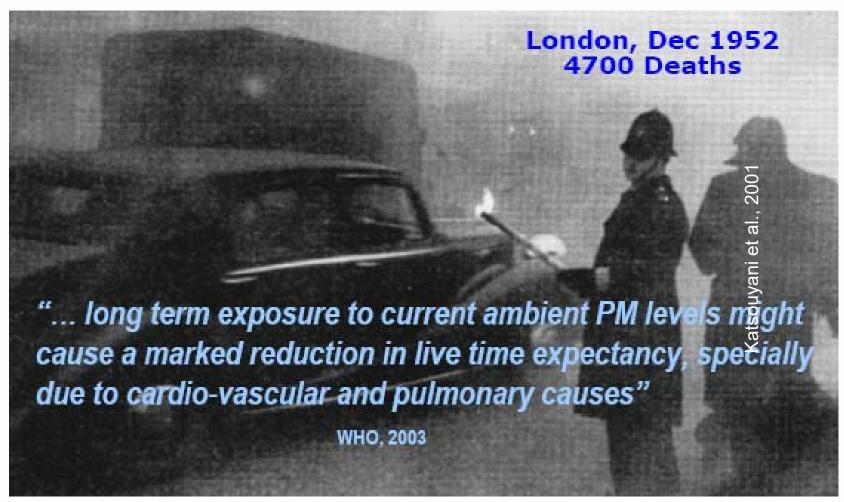


Leibniz Institute for Tropospheric Research

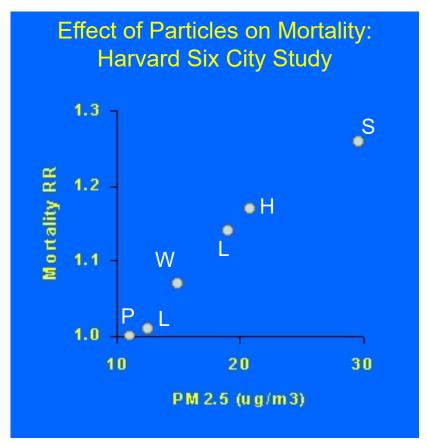
# Background

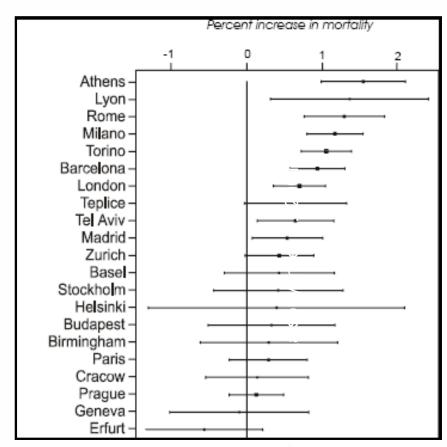
- Mass concentration measurements were the first routine measurements in networks addressing particulate ambient air pollution.
- Many epidemiological studies gave evidence that particle mass concentration is associated with adverse health effects.
- Because of these health effects particle mass concentrations are regulated (e.g. EU-Directive 2008/50/EC).
- Particle mass concentration is routinely monitored by environmental protection agencies.

# Background



# Background





Dockery et al., N. Engl. J. Med. 329, 1753 (1993)

# Particulate Matter<sub>XX</sub>

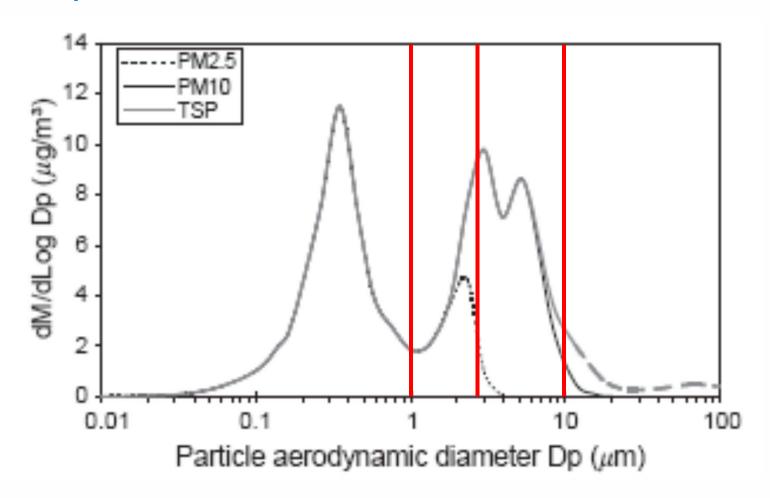
PM is often reported as particle mass concentration in  $\mu g/m^3$ , where xx is the 50% cut-off diameter of a size selective inlet

PM10, PM2.5 or PM1

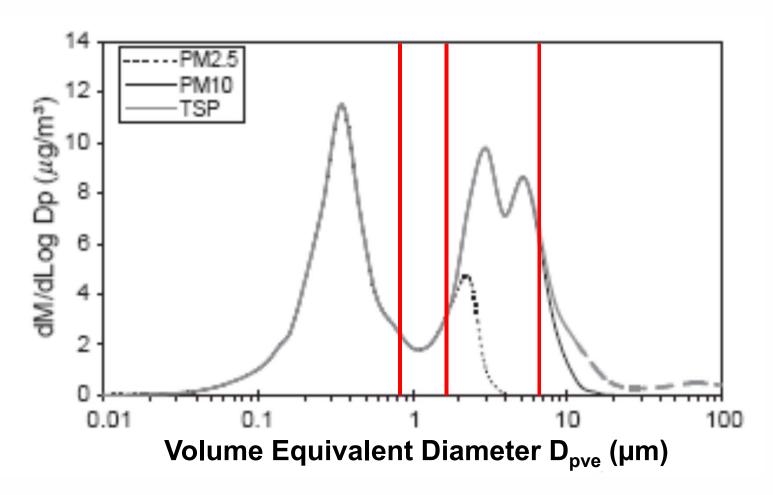
PM<sub>xx</sub> mass concentration can be determined by

- Filter sampling
- Beta- absorption
- Oscillating microbalance
- Light scattering

# **Example of Particle Mass Size Distributions**



# **Example of Particle Mass Size Distributions**



# Filter Sampling

- Filter measurements provide direct access to the aerosol property.
- Mass concentrations are simply calculated from the deposited mass on a filter and the sampled air volume.
- Standard reference method for mass concentration measurements

Mass concentration (M(loaded) – M(blank) )/Volume

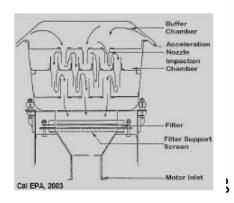
M(loaded) Mass of the loaded filter

M(blank) Mass of the blank filter prior to loading

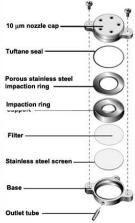
### Manual filter samplers

A wide variety of manual filter samplers is available.









# Anderson High Volume

<u>Sampler</u>

Filter size: 20x25 cm

Air flow: 1013 l/min

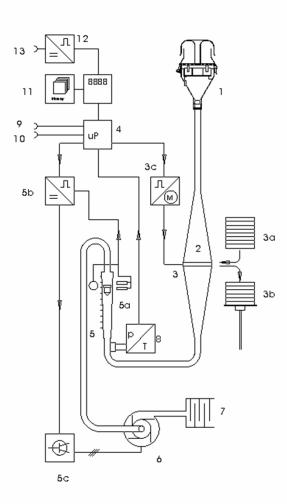
Personal Sampler

Filter diameter: 37 mm

Air flow: 2 - 10 l/min

### DIGITEL aerosol sampler DHA-80





Filter diameter: 15 cm

Air flow: 100- 1000 l/min

- 1 Pre-separator
- 2 Separator chamber
- 3 Current filter
- 3a Filter stock
- 3b Used Filter
- 3c Exchange electronics
- 4 Microprocessor control
- 5 Flow meter
- 5a Flow sensors
- 5b Flow control
- 5c Frequency converter
- 6 Blower
- 7 Noise baffle
- 8 Pressure and temperature

measurement unit

- 9 Printer Interface
- 10 RS-232C Interface
- 11 PC Card Interface
- 12 Protection lightning print

### Precision balances

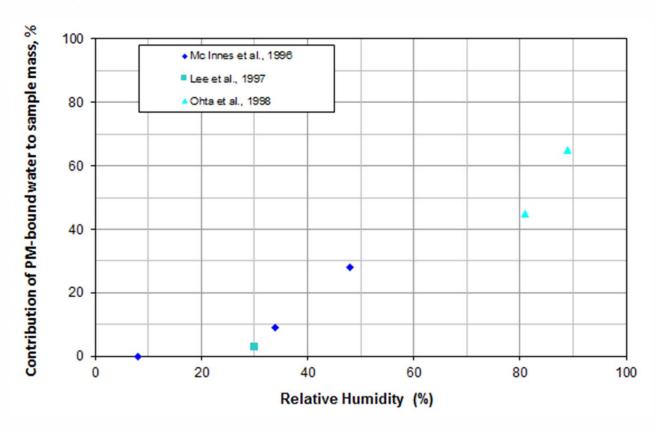




# Documents establishing the reference method

- CEN (2014) Ambient Air. Standard gravimetric measurement method for the determination of the  $PM_{10}$  or PM2,5 mass concentration of suspended particulate matter. Brussels (EN 12341).
- CEN (2017) Ambient air Automated measuring systems for the measurement of the concentration of particulate matter (PM10; PM2,5). Brussels (EN 16450).
- EC (2001) Working group on particulate matter. Guidance to member states on PM<sub>10</sub> monitoring and intercomparisons with the reference method. Draft Final Report, 16 March 2001
- EMEP (2000) EMEP-WMO Workshop on Fine Particles Emissions, Modelling and Measurements,
  Interlaken, Switzerland, 22–25 November 1999. Kjeller, EMEP/CCC-Report 9/2000
- EPA (2011) NATIONAL PRIMARY AND SECONDARY AMBIENT AIR QUALITY STANDARDS, REFERENCE METHOD FOR THE DETERMINATION OF FINE PARTICULATE MATTER AS PM2.5 IN THE ATMOSPHERE. Code of Federal Regulations, Title 40, Appendix L to part 50.
- WHO Regional Office for Europe, Copenhagen (EUR/ICP/EHB1040102, E62010, 10-13.)
- WHO (1999) Particulate Matter ( $PM_{10}$  and  $PM_{2.5}$ ). Results of Intercomparison Studies. Conference Held in Berlin 3-5 September 1998.

#### Contribution of PM-bound water



Aerosol water content expressed as mass percentage, as a function of relative humidity

Guidelines for concentration and exposure-response measurement of fine and ultra-fine particulate matter for use in epidemiological studies, EC-JRC and WHO, 2012

#### Summary gravimetric analyses

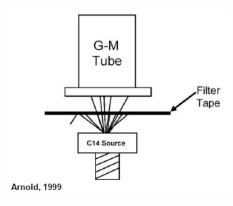
- Standard reference method
- Variety of samplers for different purposes
- Samples available for further analysis
- Reliable
- Labor-intensive
- Filter handling and weighing
- Very limited time resolution
- No online data
- Sampling artefacts
- Possible contribution of water

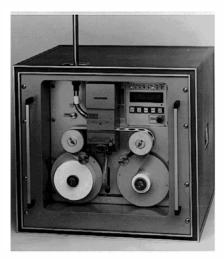


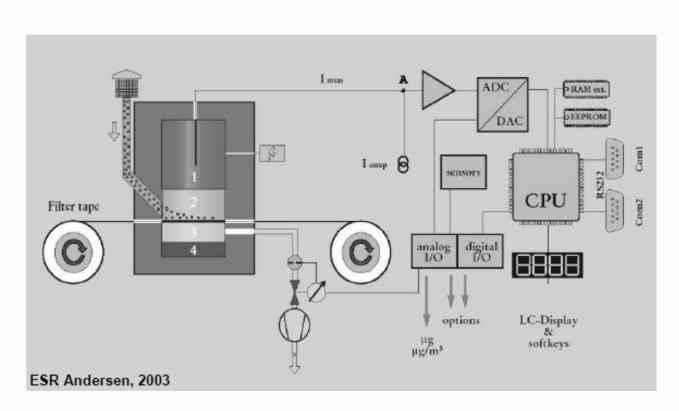


## **Beta-Absorption**

Measurement of beta-Absorption of an aerosol layer deposited on a filter







range of  $^{14}$ C b-radiation in air:  $^{\sim}$  65 cm

#### Lambert-Beer-Law

$$I = I_0 \cdot e^{-\mu \cdot x}$$

$$m = F_{\text{cal}} \cdot \ln \left( \frac{I_0}{I} \right)$$

I ... beta-Transmission of loaded filter

I<sub>0</sub> ... beta-Transmission of unloaded filter

 $\boldsymbol{\mu}$  ... Mass absorption coefficient

x ... Mass loading of the filter

### Calibration factor $F_{cal}$ is:

- proportional to the spot area A
- inverse proportional to the mass absorption coefficient  $\mu$

$$F_{\rm cal} = \frac{A}{\mu}$$

#### Mass absorption coefficient for $\beta$ radiation

$$\mu/\rho = 0.008~Z^{0.28}~E_m^{-[1.57-(Z/160)]}$$

(Katz and Penfold, 1952)

Z ... Proton number

A ... Mass number

 $\mu/\rho = 15.2 Z^{(4/3)} A^{-1} E_m^{-1.485}$ 

(Thummel, 1974)

E<sub>m</sub> ... Maximum electron energy

#### Baron/Willeke, 2001

Compound	Z/A	$\mu  (\mathrm{cm}^2/\mathrm{mg})^a$	$\mu  (\text{cm}^2/\text{mg})^b$
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	0.530	0.153	0.166
NH4HSO4	0.521	0.152	0.163
CaSO <sub>4</sub> · 2H <sub>2</sub> O	0.511	0.152	0.159
SiO <sub>2</sub>	0.499	0.154	0.154
CaCO <sub>3</sub>	0.500	0.154	0.154
Carbon	0.500	0.154	0.154
Fe <sub>2</sub> O <sub>3</sub>	0.476	0.163	0.144
NaCl	0.478	0.172	0.145
PbSO <sub>4</sub>	0.429	0.193	0.126
PbCl <sub>2</sub>	0.417	0.204	0.121
PbBrCl	0.415	0.206	0.120

- a) experimentally determined
- b) calculated

The mass absorption coefficient  $\mu$  is material dependent.





- The mass load of the filter is measured in time intervals  $\tau$ .
- The change of filter mass is calculated as:

$$\frac{\mathrm{d}m}{\mathrm{d}t} = \frac{\sum (t_i - \overline{t}) \cdot m_i}{\sum (t_i - \overline{t}) \cdot t_i} = \frac{\Delta m}{\tau}$$

t<sub>i</sub> ... measurement timet ... average of measurement timesm<sub>i</sub> ... mass measured at measurement

time

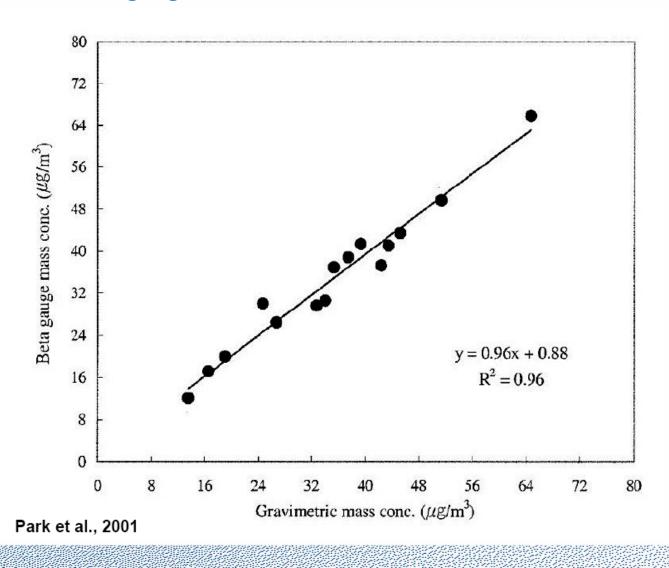
The mass concentration *C* is calculated knowing the average volumetric flow rate *Q*:

$$C = \frac{\mathrm{d}m / \mathrm{d}t}{Q}$$





### Co-located filter and beta-gauge measurements



#### **Summary Beta-Absorption**

#### **Advantages**

- Online measurement
- Long-term stability up to one year
- Good agreement with filter samplers
- US-EPA approved reference method

#### **Disadvantages**

- Moderate dependency on chemical composition of aerosol particles
- Filter typically not usable for chemical analysis
- Radioactive source may require a permit in some countries

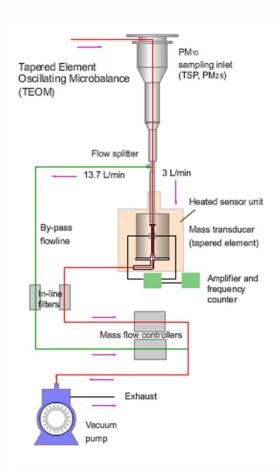


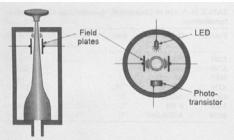


# Tapered Element Oscillating Microbalance (TEOM)

Frequency change of an oscillating filter with increasing mass load









The tapered element with the filter on top behaves like an oscillating pendulum.

$$f^2 = \frac{K_0}{m}$$

f ... Eigen-frequency of the tapered element with filter

K<sub>0</sub> ... Calibration constant

m ... Mass

$$m = m_F + m_0 + \delta m$$

m<sub>F</sub> ... Mass of the blank filter

m<sub>0</sub> ... Mass of the tapered element

 $\delta$ m... Deposited mass on the filter

The frequency of the pendulum decreases with increasing mass deposited on the filter.

An increase of the filter mass by  $\delta m$  causes a frequency change from  $f_i$  to  $f_f$ 

$$f_i^2 = \frac{K_0}{m_F + m_0}$$

$$f_f^2 = \frac{K_0}{m_F + m_0 + \delta m}$$

Mass change can be calculated from the frequency change

$$\delta m = \mathbf{K}_0 \cdot \left( \frac{1}{f_f^2} - \frac{1}{f_i^2} \right)$$

And the mass concentration can be calculated to

$$C = \frac{\delta m \cdot \Delta t}{Q}$$

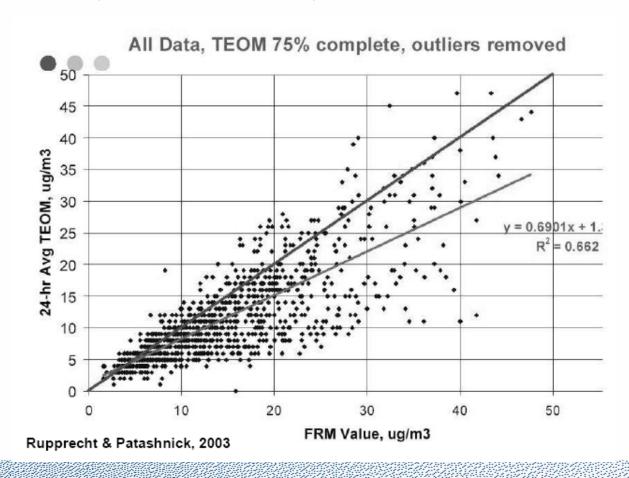
 $\Delta t...$  Time from  $f_i$  to  $f_f$ 

Q... Volumetric flow rate

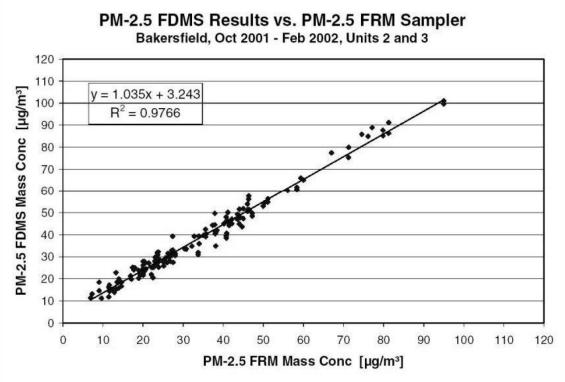




- Diurnal variation of RH causes variation of filter weight.
- Standard TEOM uses heated inlet at factory setting 35°C.
- Ammonium nitrate evaporates at this temperature.



- Reduce temperature to 30°C
- Remove water with Nafion dryer (TEOM-SES)
- Alternating measurement between filtered air and aerosol (differential TEOM, TEOM-FDMS)









### **Summary TEOM**

### Advantages

- Online measurement
- Long-term stability up to several months
- Good agreement with filter samplers if used with appropriate RH control or correction
- US-EPA approved reference method at 30°C

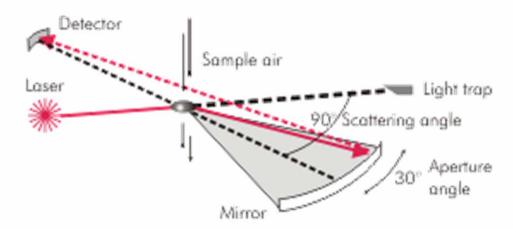
### Disadvantages

 Standard TEOM measurements are influenced by chemical composition of the aerosol particles if used according to factory settings

# **Optical Instruments**

- Measurement of light scattered by single aerosol particles in a measurement volume under a fixed angle
- Sizing by pulse height analysis





- The instrument is calibrated with well defined aerosol particles such as Latex particles
- Direct indication of particle mass concentration
- Light scattered from a measurement volume depends on many aerosol properties
  - Particle shape
  - Particle size distribution
  - Refractive index
- Particle density is assumed for conversion to mass concentration
- Light weight, inexpensive monitor
- The instrument is calibrated against reference